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Key indicators

Single-crystal X-ray study T = 213 KMean $\sigma(\text{F-B}) = 0.002 \text{ Å}$ Disorder in main residue R factor = 0.028 wR factor = 0.084 Data-to-parameter ratio = 12.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. Cobaltocenium tetrafluoroborate, $[Co(C_5H_5)_2]BF_4$, crystallizes in space group *Cccm* and packs in a way that causes the Cp₂Co moiety to sit just off of the mirror plane, causing extreme disorder.

A highly disordered cobaltocenium salt

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Comment

Cobaltocene has often been used as a convenient, mild reducing agent (Connelly & Geiger, 1996), thus leading to the appearance of the cobaltocenium cation in many crystal structures [over 80 times in the Cambridge Database (Allen & Kennard, 1993)]. We have also found the cobaltocenium cation to be useful as a mild oxidizing agent, and often use the BF_4^- salt whenever a weakly coordinating anion is desired in the target synthetic product. While structures of the Cp_2Co^+ ion with many different anions are known, the BF_4^- salt, (I), is not, and its structure is reported here.



Space group Cccm was strongly indicated by intensity statistics, and while the solution and refinement of the structure was expected to be straightforward, the initial solution from direct methods suggested a position for the Co atom that was on the twofold axis at 1/4, 1/4, z but only approximately 0.2 Å from the 2/m site at 1/4, 1/4, 1/2. (The BF₄⁻ anion resides on a site of 222 symmetry at 0, 1/2, 1/4 in a way that generates an almost perfectly tetrahedral moiety.) Fourier peaks corresponding to the C atoms of the Cp rings were therefore badly mixed by the symmetry operations. At this point, it seemed apparent that the mirror plane should be removed to reduce the symmetry and change the space group to Ccc2. However, two slightly separated positions for the Co atom persisted and the Cp rings continued to appear as a badly disordered cluster of atoms. The solution and refinement were therefore continued in space group Cccm, using a disordered model in which the Cp_2Co^+ unit resides on a twofold axis perpendicular to the main axis of the molecule, but slightly displaced from the mirror plane which almost contains the main molecular axis. A view of the model showing one orientation of the Cp₂Co group is shown in Fig. 1.

In order to refine the model effectively, the Cp ring was constrained to be exactly pentagonal, but was allowed to

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metal-organic papers



Figure 1

The molecular structure of [(C₅H₅)₂Co]BF₄, drawn with 40% displacement ellipsoids.

shrink or expand freely. This resulted in a C-C distance of 1.397 (4) Å, which is shorter than the expected distance for π bonded Cp, but is due to librational error. Librational analysis (Schomaker & Trueblood, 1968) of the Cp ring gives corrected C-C bond lengths of 1.42 Å, which is the expected value of the distances.

Atom C1 lies almost on the mirror plane and appears to be shared by both orientations (see Fig. 2), although it is actually not. The Cp plane is 1.634 (1) Å from the Co atom, and the two Cp rings form an angle of $1.5 (4)^{\circ}$. The Cp rings are slightly staggered, with an approximate offset of 21° .

Experimental

All synthetic procedures were carried out under a nitrogen atmosphere. Cobaltocene (51 mg, 0.27 mmol) and AgBF₄ (52 mg, 0.27 mmol) were weighed into separate Schlenk flasks. Methylene chloride (10 ml) was added to each flask, and the cobaltocene solution was transferred to the flask containing the AgBF₄ via cannula. The mixture was allowed to react, with stirring, for 1 h. The solution was then transferred with a filter-equipped cannula into a 50 ml Schlenk tube and layered with 20 ml hexanes. Crystals formed after one week.

Crystal data

$[Co(C_{5}H_{5})_{2}]\cdot BF_{4}$ $M_{r} = 275.92$ Orthorhombic, <i>Cccm</i> $a = 7.3586$ (7) Å $b = 11.9176$ (11) Å $c = 12.4369$ (13) Å $V = 1090.68$ (18) Å ³ $Z = 4$ $D_{x} = 1.680 \text{ Mg m}^{-3}$ Defense M_{2} and M_{2}	Mo K α radiation Cell parameters from 1619 reflections $\theta = 2.4-27.4^{\circ}$ $\mu = 1.59 \text{ mm}^{-1}$ T = 213 (2) K Block, yellow $0.30 \times 0.28 \times 0.20 \text{ mm}$
Bruker SMART diffractometer ω scans Absorption correction: multi-scan (<i>SADABS</i> ; Blessing, 1995) $T_{min} = 0.609, T_{max} = 0.727$ 3263 measured reflections 665 independent reflections	485 reflections with $I > 2\sigma(I)$ $R_{int} = 0.018$ $\theta_{max} = 27.5^{\circ}$ $h = -9 \rightarrow 9$ $k = -15 \rightarrow 10$ $l = -16 \rightarrow 13$



Figure 2

A representation of the entire model used in the refinement of the disordered structure. The twofold axis is horizontal in this view, passing through the Co atoms. The mirror is vertical, perpendicular to the plane of the drawing, passing between the Co atoms.

Refinement
Refinement on F^2
$R[F^2 > 2\sigma(F^2)] = 0.028$
$wR(F^2) = 0.084$
S = 1.09
665 reflections
55 parameters
H-atom parameters not refined

 $w = 1/[\sigma^2(F_o^2) + (0.036P)^2]$ + 0.6P] where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\rm max} = 0.003$ $\Delta \rho_{\rm max} = 0.19 \ {\rm e} \ {\rm \AA}^{-3}$ $\Delta \rho_{\rm min} = -0.18 \text{ e } \text{\AA}^{-3}$

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Siemens, 1996); software used to prepare material for publication: SHELXL97.

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