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## F. Albert Cotton,* Lee M. Daniels and Chad C. Wilkinson

Laboratory for Molecular Structure and Bonding, Dept. of Chemistry, Texas A\&M University,
PO Box 30012, College Station,
TX 77842-3012, USA
Correspondence e-mail: cotton@tamu.edu

## Key indicators

Single-crystal X-ray study
$T=213 \mathrm{~K}$
Mean $\sigma(\mathrm{F}-\mathrm{B})=0.002 \AA$
Disorder in main residue
$R$ factor $=0.028$
$w R$ factor $=0.084$
Data-to-parameter ratio $=12.1$

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## A highly disordered cobaltocenium salt

Cobaltocenium tetrafluoroborate, $\left[\mathrm{Co}\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)_{2}\right] \mathrm{BF}_{4}$, crystallizes in space group Cccm and packs in a way that causes the $\mathrm{Cp}_{2} \mathrm{Co}$ moiety to sit just off of the mirror plane, causing extreme disorder.

## Comment

Cobaltocene has often been used as a convenient, mild reducing agent (Connelly \& Geiger, 1996), thus leading to the appearance of the cobaltocenium cation in many crystal structures [over 80 times in the Cambridge Database (Allen \& Kennard, 1993)]. We have also found the cobaltocenium cation to be useful as a mild oxidizing agent, and often use the $\mathrm{BF}_{4}^{-}$salt whenever a weakly coordinating anion is desired in the target synthetic product. While structures of the $\mathrm{Cp}_{2} \mathrm{Co}^{+}$ ion with many different anions are known, the $\mathrm{BF}_{4}{ }^{-}$salt, (I), is not, and its structure is reported here.

$\mathrm{BF}_{4}{ }^{-}$
(I)

Space group Cccm was strongly indicated by intensity statistics, and while the solution and refinement of the structure was expected to be straightforward, the initial solution from direct methods suggested a position for the Co atom that was on the twofold axis at $1 / 4,1 / 4, z$ but only approximately $0.2 \AA$ from the $2 / m$ site at $1 / 4,1 / 4,1 / 2$. (The $\mathrm{BF}_{4}{ }^{-}$anion resides on a site of 222 symmetry at $0,1 / 2,1 / 4$ in a way that generates an almost perfectly tetrahedral moiety.) Fourier peaks corresponding to the C atoms of the Cp rings were therefore badly mixed by the symmetry operations. At this point, it seemed apparent that the mirror plane should be removed to reduce the symmetry and change the space group to $C c c 2$. However, two slightly separated positions for the Co atom persisted and the Cp rings continued to appear as a badly disordered cluster of atoms. The solution and refinement were therefore continued in space group Cccm, using a disordered model in which the $\mathrm{Cp}_{2} \mathrm{Co}^{+}$unit resides on a twofold axis perpendicular to the main axis of the molecule, but slightly displaced from the mirror plane which almost contains the main molecular axis. A view of the model showing one orientation of the $\mathrm{Cp}_{2}$ Co group is shown in Fig. 1.

In order to refine the model effectively, the Cp ring was constrained to be exactly pentagonal, but was allowed to

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Figure 1
The molecular structure of $\left[\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)_{2} \mathrm{Co}\right] \mathrm{BF}_{4}$, drawn with $40 \%$ displacement ellipsoids.
shrink or expand freely. This resulted in a $\mathrm{C}-\mathrm{C}$ distance of 1.397 (4) $\AA$, which is shorter than the expected distance for $\pi$ bonded Cp, but is due to librational error. Librational analysis (Schomaker \& Trueblood, 1968) of the Cp ring gives corrected C-C bond lengths of $1.42 \AA$, which is the expected value of the distances.

Atom C1 lies almost on the mirror plane and appears to be shared by both orientations (see Fig. 2), although it is actually not. The Cp plane is 1.634 (1) $\AA$ from the Co atom, and the two Cp rings form an angle of $1.5(4)^{\circ}$. The Cp rings are slightly staggered, with an approximate offset of $21^{\circ}$.

## Experimental

All synthetic procedures were carried out under a nitrogen atmosphere. Cobaltocene ( $51 \mathrm{mg}, 0.27 \mathrm{mmol}$ ) and $\mathrm{AgBF}_{4}(52 \mathrm{mg}$, $0.27 \mathrm{mmol})$ were weighed into separate Schlenk flasks. Methylene chloride ( 10 ml ) was added to each flask, and the cobaltocene solution was transferred to the flask containing the $\mathrm{AgBF}_{4}$ via cannula. The mixture was allowed to react, with stirring, for 1 h . The solution was then transferred with a filter-equipped cannula into a 50 ml Schlenk tube and layered with 20 ml hexanes. Crystals formed after one week.

## Crystal data

$\left[\mathrm{Co}\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)_{2}\right] \cdot \mathrm{BF}_{4}$
$M_{r}=275.92$
Orthorhombic, Cccm
$a=7.3586$ (7) $\AA$
$b=11.9176$ (11) $\AA$
$c=12.4369$ (13) $\AA$
$V=1090.68(18) \AA^{3}$
$Z=4$
$D_{x}=1.680 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 1619
reflections
$\theta=2.4-27.4^{\circ}$
$\mu=1.59 \mathrm{~mm}^{-1}$
$T=213$ (2) K
Block, yellow
$0.30 \times 0.28 \times 0.20 \mathrm{~mm}$

## Data collection

Bruker SMART diffractometer

## $\omega$ scans

Absorption correction: multi-scan
(SADABS; Blessing, 1995)
$T_{\text {min }}=0.609, T_{\text {max }}=0.727$
3263 measured reflections
665 independent reflections


Figure 2
A representation of the entire model used in the refinement of the disordered structure. The twofold axis is horizontal in this view, passing through the Co atoms. The mirror is vertical, perpendicular to the plane of the drawing, passing between the Co atoms.

## Refinement

Refinement on $F^{2}$
Refinement
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.028$
$w R\left(F^{2}\right)=0.084$
$S=1.09$
665 reflections
55 parameters
H -atom parameters not refined

$$
\begin{aligned}
& w=1 /[ \sigma^{2}\left(F_{o}^{2}\right)+(0.036 P)^{2} \\
&+0.6 P] \\
& \text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=0.003 \\
& \Delta \rho_{\max }=0.19 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.18 \mathrm{e} \AA^{-3}
\end{aligned}
$$

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Siemens, 1996); software used to prepare material for publication: SHELXL97.

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